

Sensitive Thermography via Sensing Visible Photons Detected from the Manipulation of the Trap State in MAPbX₃

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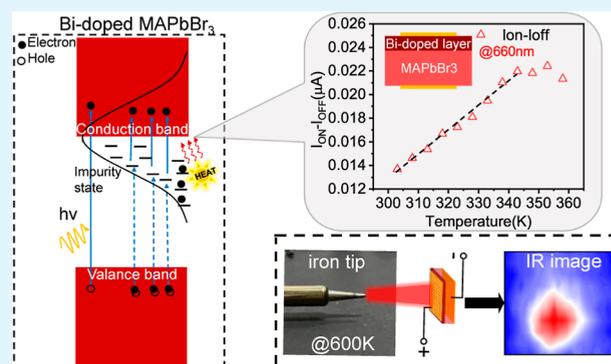
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ABSTRACT: Sensitive thermometry or thermography by responding to blackbody radiation is urgently desired in the intelligent information life, including scientific research, medical diagnosis, remote sensing, defense, etc. Even though thermography techniques based on infrared sensing have undergone unprecedented development, the poor compatibility with common optical components and the high diffraction limit impose an impediment to their integration into the established photonic integrated circuit or the realization of high-spatial-resolution and high-thermal-resolution imaging. In this work, we present a sensitive temperature-dependent visible photon detection in Bi-doped MAPbX₃ (X = Cl, Br, and I) and employ it for uncooled thermography. Systematic measurements reveal that the Bi dopant introduces trap states in MAPbX₃, thermal energy facilitates the carriers jumping from trap states to the conduction band, while the vacancies of trap states ensure the sequential absorption of visible photons with energy less than the band gap. Subsequently, the change of response toward the visible photon is applied to construct the thermograph, and it possesses a specific sensitivity of 2.11% K⁻¹ along temperature variation. As a result, our thermograph presents a temperature resolution of 0.21 nA K⁻¹, a high responsivity of 2.06 mA W⁻¹, and a high detectivity of 2.08 × 10⁹ Jones at room temperature. Furthermore, remote thermal imaging is successfully achieved with our thermograph.

KEYWORDS: thermograph, photodetector, perovskite, Bi-MAPbBr₃, single crystal



INTRODUCTION

Beyond visible or near-infrared photons, the acquisition of thermal radiation from the target can provide more identifiable details under weak visible light conditions disregarding disturbance from the normal background.^{1,2} Benefiting from this merit, until now, thermal sensing techniques have been intensely explored in many fields, for example, night vision,³ medical science,^{4,5} electronic systems,⁶ security,⁷ and artificial intelligence.⁸ Especially, as the electric car and computational power undergo extraordinary splendor in recent years, thermal detection is showing increasing importance in their intelligent safety systems for ensuring the safety in battery packs, server clusters, etc.⁹ Even though it is well-known that the thermal radiation is an infrared (IR) electromagnetic wave and the detection of IR photons can be used to acquire thermal information since the temperature variation can induce the change in IR emission according to the blackbody radiation law

$$B(T) = \sigma T^4 \quad (1)$$

where σ is the Stefan–Boltzmann constant and T is the temperature,^{10,11} the detection of the radiative IR photon still faces a challenge due to its feeble energy.

In general, the technique for the realization of thermography can be categorized into two types: photoelectric effect and photothermal effect.^{1,12} For the photoelectric effect, it uses the photodetectors with a narrow-bandgap semiconductor (InSb, In_{1-x}Ga_xAs, and Hg_{1-x}Cd_xTe) or a GaAs/AlGaAs quantum well to respond to the IR photon radiation.^{13,14} In this case, the photodetectors convert the IR photon into the electric signal directly with a high response speed. However, because the band gap of active materials is even lower than that of thermal energy at room temperature, the detection of IR photons under room temperature requires additional cryogenic cooling

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units to suppress the air thermal energy pumping the carrier to the conduction band, which inevitably increases the cost and complexity in devices and subsequently imposes an impediment for its universal application. As a consequence, thermography with an IR photodetector is mainly used in scientific or military equipment.

Alternatively, the photothermal effect including bolometric, thermoelectric, and pyroelectric effects provides another option for constructing the affordable thermograph.¹⁵ In this approach, the thermal sensing materials absorb IR radiation and convert its thermal energy into detectable signals, including electric or photon signals. For instance, Jiang et al. reported conductivity–temperature dependence in GeTe,¹⁶ Reihani et al. reported that GaAs exhibits temperature-dependent reflectance at the chosen wavelength,¹⁷ and our previous work confirms the conductivity–temperature dependence in perovskite materials.¹⁸ All of these attempts demonstrate success in thermal detection. However, it is worth noting that although the photothermal effect empowers the thermal imagery without cooling equipment, it suffers from a slow response speed.

Ultimately, the most important technical challenge facing in both photoelectric effect- and photothermal effect-based thermography is that they cannot be compatible with the conventional optical system due to the poor transmission of thermal energy through the optical component.^{17,19} Notably, their integration into the established optical system is extremely essential in realizing high-resolution imaging. Hence, together with their severe diffraction, the thermographic performance is still far from satisfactory due to the limitation in spatial resolution and thermal resolution. Thus, the technique of tracing the change in fluorescence or its lifetimes has been exploited and regarded as a potential way to achieve high-performance nanoscale thermography.

Profiting from the prosperity of development in novel temperature-sensitive materials, Neumann and his co-authors used single defects in diamonds to detect small and local temperature changes.²⁰ Yakunin et al. presented the strong temperature dependence of the luminescence lifetime of perovskite-like tin halides.¹ Besides, some other perovskites, especially those exhibiting thermoelectric characteristics, their application in thermal detection have also been successfully demonstrated.^{21,22} Even so, the stringent requirement on the complicated measurement system still imposes restrictions on their practical application. So, developing feasible and sensitive thermography is still a great desire and challenge for the scientific, industrial, and even military fields.

In this work, we report sensitive thermography with Bi-doped MAPbX₃ (Bi-MAPbX₃) as a temperature-active material. The dopant of Bi in MAPbX₃ induces the traps, resulting in the extra absorption of photons with energy less than the optical band gap. Moreover, the trapped carriers can be pumped to the conduction band by the thermal energy. Thus, the trap and detrap processes lead to the change in the absorption of the tailing photon at different temperatures, which enables the potential in thermal detection with the common photodetector mode. To construct this thermograph and boost its performance, a p–n junction with Bi-MAPbX₃ on MAPbX₃ was fabricated to enhance the charge carrier transport and collection with Au as electrodes. As a result, our thermograph by responding to visible photons shows an excellent temperature sensitivity of 2.11% K⁻¹, a high thermal resolution of 0.21 nA T⁻¹, a high responsivity of 0.23 mA W⁻¹,

and long-term stability. With the assistance of our thermograph, remote thermal imaging of soldering iron with >2 μm IR wave radiation is successfully achieved.

EXPERIMENTAL SECTION

Materials. Lead chloride (PbCl₂, 99%) and lead bromine (PbBr₂, 98%) were purchased from Sigma-Aldrich, USA. Methylammonium chloride (MACl) and methylammonium bromide (MABr) were purchased from Sigma-Aldrich, USA.

Bismuth chloride (BiCl₃, 99.9%) and bismuth bromide (BiBr₃, 99%) were purchased from Sigma-Aldrich, USA. Dimethyl sulfoxide (DMSO) and dimethylformamide-*d*₇ (DMF) were obtained from Aladdin, China. All commercial products were used as received.

Precursor Preparation. For intrinsic MAPbCl₃ precursor solution preparation, 1 mol L⁻¹ MACl and 1 mol L⁻¹ PbCl₂ were dissolved in the mixed solution of 50 mL of DMF and 10 mL of DMSO. For intrinsic MAPbBr₃ precursor solution preparation, 1 mol L⁻¹ MABr and 1 mol L⁻¹ PbBr₂ were dissolved in 60 mL of DMF. Additionally, to obtain the Bi-doped perovskite precursor, 0.1 mol L⁻¹ BiCl₃ and 0.1 mol L⁻¹ BiBr₃ were added into the intrinsic MAPbCl₃ and MAPbBr₃ perovskite precursor solutions, respectively, which were stirred at room temperature for 2 h.

Device Fabrication. The inverse temperature crystallization (ITC) method was used here to grow Bi-doped and intrinsic MAPbCl₃ and MAPbBr₃ single crystals (SCs). The intrinsic MAPbCl₃ precursor solution and Bi-MAPbCl₃ precursor solution were heated from 40 to 60 °C at the speed of 1 °C per 3 h to grow SCs.

Additionally, the intrinsic MAPbBr₃ precursor solution and Bi-MAPbBr₃ precursor solution were heated from 60 to 80 °C at the speed of 1 °C per 3 h to grow crystals. The solution-processed epitaxial (SPE) growth method was used to grow Bi-MAPbBr₃ layers by digging intrinsic MAPbBr₃ SCs into the precursor with the Bi-MAPbBr₃ precursor solution. After the SPE growth process, the perovskite SCs with a core–shell structure were cut by a diamond wire (Φ 0.35 mm) and were polished by diamond powder (Φ 0.5 μm) and toluene mixed solution. The cutting and polishing machines (STX-202A and UNIPOL-1203) were obtained from Shenyang Kejing Autoinstrument Co., Ltd. (Shenyang, China). What is more, the gold electrodes were deposited on the opposite surface of perovskite SCs in vacuum.

Material Characterization. The samples were held in the larger chamber of INSTEC's HCS402 (USA), which has dual top–bottom heating for reduced vertical temperature gradients, and the high precision thermal control was realized by INSTEC mK2000 (USA). The temperature-dependent absorption of Bi-doped perovskite SCs was measured by a monochromator (Omin-λ200i, Zolix) and a silicon PIN photodiode (S12158-01CT, Hamamatsu) with power supplied by Keithley 2400. The photoluminescence (PL) decay processes were characterized with a SpectraMax instrument (UK). X-ray photoelectron spectroscopy (XPS) was performed by a PHI 5000 Versa Probe (Japan), and X-ray diffraction (XRD) patterns were obtained by an X'TRA system with a Cu target (Switzerland). The contact potential difference was measured by a Kelvin probe force microscopy (KPFM) system from Multimode-8-AM (USA), and the major carrier density was measured by a Hall effect measurement system (PM-50 MR Platform, EMT-2400 Data Acquisition System).

Device Measurement. Scanning electron microscopy (SEM) images were taken with a Quanta 200 FEI (USA), transmission electron microscopy (TEM) images were obtained by a Titan 80-300 FEI (USA), and high-resolution TEM (HR-TEM) images were obtained by a JEM-200CX (JAPAN). The pulsed light of 660 nm was provided by a light-emitting diode (L3882, Hamamatsu) powered by a signal generator (33600A, KEYSIGHT). The laser of 1550 nm was obtained by MDL-N-1550-3W (Changchun, China).

THEORETICAL CALCULATIONS

Density Functional Theory Calculation. Density functional theory (DFT) calculations were conducted using the

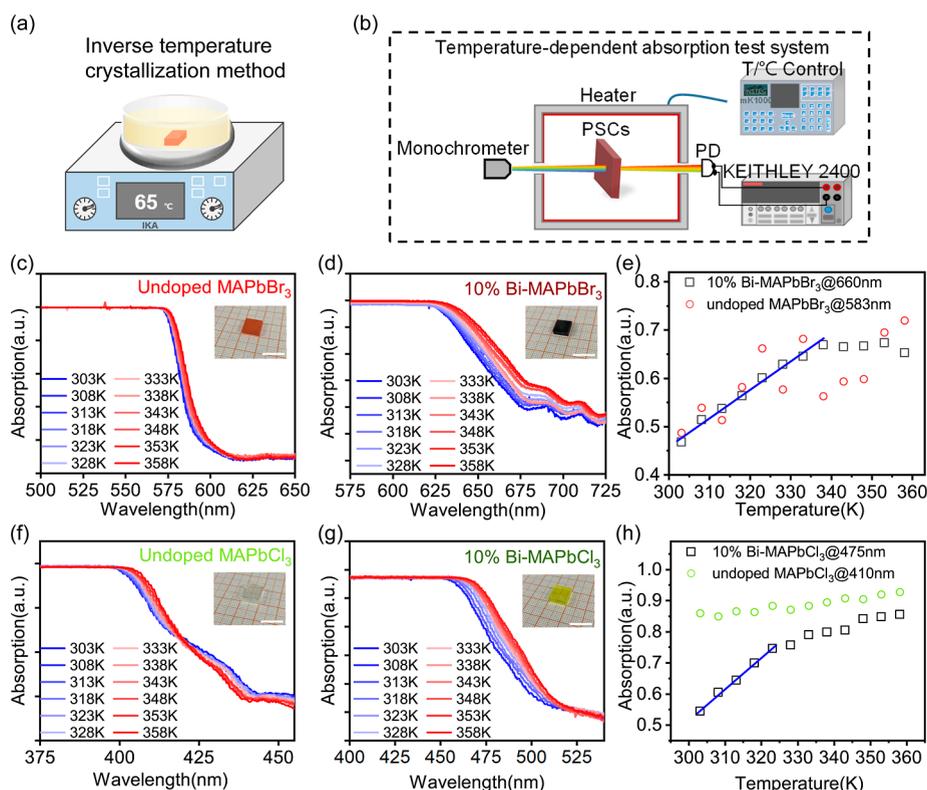


Figure 1. Temperature-dependent absorption characterization of perovskite SCs. (a) Growth process of SCs by the ITC method. (b) Schematic diagram of the temperature-dependent absorption test system. The PD is based on a silicon PIN detector. (c) Normalized absorption spectra of undoped MAPbBr₃ SCs at various temperatures. (d) Normalized absorption spectra of 10% Bi-MAPbBr₃ SCs at various temperatures. (e) Absorptivity temperature dependence for undoped MAPbBr₃ SCs (@583 nm) and 10% Bi-MAPbBr₃ SCs (@660 nm) at their most thermal sensitive wavelength. (f) Normalized absorption spectra of undoped MAPbCl₃ SCs at various temperatures. (g) Normalized absorption spectra of 10% Bi-MAPbCl₃ SCs at various temperatures. (h) Absorptivity temperature dependence for undoped MAPbCl₃ SCs (@410 nm) and 10% Bi-MAPbCl₃ SCs (@475 nm) at their most thermal sensitive wavelength.

projector-augmented wave method as implemented in the VASP code.²³ The plane-wave cutoff energy was set to 450 eV. The generalized gradient approximation (GGA) Perdew–Burke–Ernzerhof (PBE) functional and Γ -centered grid k -meshes with a k -spacing of 0.3 Å⁻¹ were employed for structural relaxations, which stopped when the total force on each atom was smaller than 0.03 eV/Å.

RESULTS AND DISCUSSION

According to the blackbody radiation law, objects at a temperature above absolute zero can emit IR radiation, which is an electromagnetic wave that cannot be directly seen by human eyes.⁷ Therefore, the thermograph by responding to radiative IR photons shows significance in identifying the substances under the weak visible light condition, such as monitoring the battery safety, chip temperature, agricultural production,²⁴ and human body health,⁴ as depicted in Figure S1. However, their poor integration in the established optical system imposes an impediment to their practical application due to the poor transmission of thermal energy. Thus, in this work, we will develop a sensitive and feasible thermograph by sensing the visible photons.

Doping is a common method for inducing defects or excitation in materials and shows the influence on their intrinsic properties including the optical band gap, carrier transport, or recombination.²⁵ The manipulation of the doping concentration was designed to form a doping level for realizing

IR detection.²⁶ Taking the star material of lead halide perovskites as an example, it undergoes an explosive development in optoelectronic devices due to the merits in the facile solution processing fabrication and excellent optoelectronic properties.²⁷ The soft lattice in the perovskite also promises the lead analogous ion dopant incorporation.²⁸ Until now, the heterovalent dopants of Ag, Sb, Bi, Fe, Ce, Sr, etc. have been explored as dopants, in which Bi can be used as an n-type dopant and the doping concentration can be easily controlled with the experiment condition.²⁹ Moreover, it is revealed that the Bi substitution in the perovskite lattice can significantly modify its thermoelectric properties.²² Thus, in this work, the Bi-MAPbX₃ SC is selected as the sensitive material.

Figure 1a shows the schematic diagram of solution processing fabrication for the proposed Bi-MAPbX₃ SCs, in which the ITC method is used.³⁰ To facilitate the doping of Bi into MAPbX₃, beyond the stoichiometric ratio between PbX₂ and MAX, extra BiX₃ was added to the precursor.

The details can be found in the Experimental Section. To identify whether the doping is beneficial for the thermography, the temperature-dependent absorption analysis in MAPbBr₃ and Bi-MAPbBr₃ was conducted. Figure 1b depicts a schematic diagram of the homemade testing system. Similar to spectrometers, this system consists of a monochromator, a photodetector, and a microcomputer for recording spectra. The heater and the precision temperature controller are essential parts in the temperature-dependent absorption test. During the measurement, the SCs are held in the sample

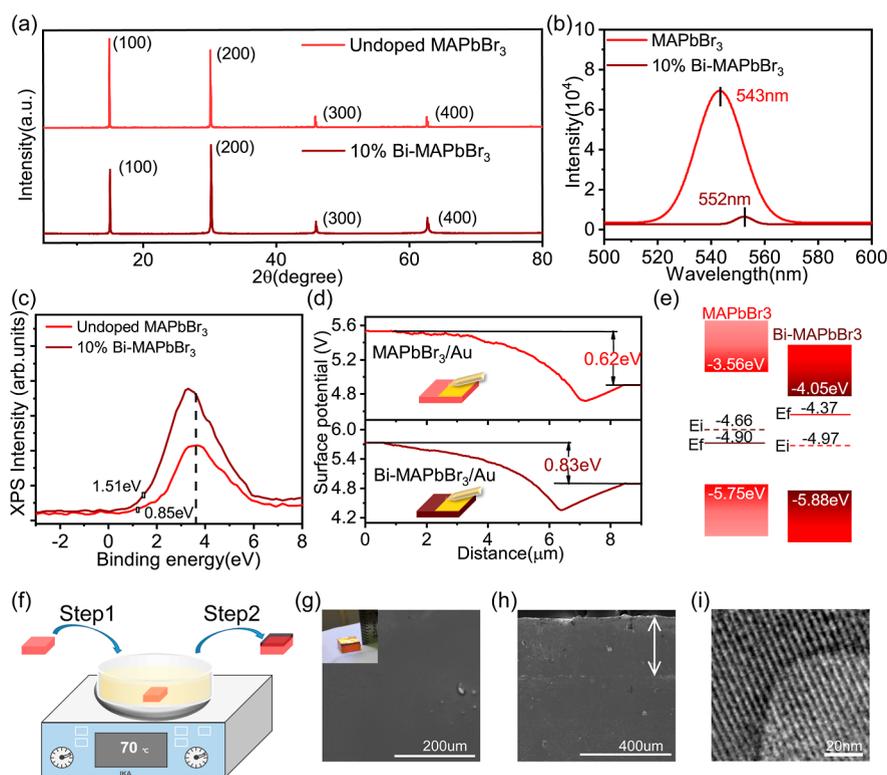


Figure 2. Basic optical properties of undoped MAPbBr₃, Bi-MAPbBr₃, and the epitaxial homostructure. (a) XRD patterns of undoped MAPbBr₃ SCs and 10% Bi-MAPbBr₃ SCs, respectively. (b) PL spectra of undoped MAPbBr₃ SCs and Bi-MAPbBr₃ SCs, respectively. (c) XPS spectra of undoped MAPbBr₃ SCs and 10% Bi-MAPbBr₃ SCs, respectively. (d) KPFM measurements of undoped and 10% Bi SCs and surface potential differences between undoped MAPbBr₃ SCs and gold and 10% Bi-MAPbBr₃ SCs and gold, respectively. (e) Band gap alignment of undoped MAPbBr₃ and 10% Bi-MAPbBr₃. (f) Schematic diagram of the SPE growth process. (g) SEM image of the surface of the epitaxial Bi-MAPbBr₃ layer. The scale bar is 200 μ m. The inset is the picture of the MAPbBr₃ substrate with the epitaxial Bi-MAPbBr₃ layer. (h) SEM image of the cross-section of the Bi-MAPbBr₃ and undoped MAPbBr₃ homostructure. The scale bar is 400 μ m. (i) HRTEM image of the interface between the substrate and epitaxial layer. The scale bar is 20 nm.

chamber with a dual top–bottom heater, which virtually eliminates vertical temperature gradients in the sample chamber. The top and bottom dual pane windows are used for visible photons going through the sample chamber. Figure 1c,d shows the normalized absorption spectra in MAPbBr₃ SCs and 10% Bi-MAPbBr₃ SCs at various temperatures. The inset shows the photograph of SCs. As expected, a significant enhancement is observed in the absorption edge with the increasing temperature.

Compared with MAPbBr₃ SCs, 10% Bi-MAPbBr₃ SCs possess a clear red shift in absorption, and the energy of the photon is even beyond the optical band gap. This result further verifies that the Bi ions incorporation forms the interband, which is essential for low-energy IR photon detection or constructing the thermograph. More importantly, the variation of visible photon absorption along with the change in the temperature also lays a foundation for realizing sensitive thermography. To magnify this thermal sensitivity, Figure 1e presents the absorption toward the characteristic photons of 660 and 583 nm in Bi-MAPbBr₃ and MAPbBr₃ SCs, respectively.

Although both Bi-MAPbBr₃ and MAPbBr₃ present the obvious change, only the absorption in Bi-MAPbBr₃ shows a linear variation and the linear dynamic region ranges from 303 to 343 K. Notably, this linear variation is important in enabling the high thermal resolution in temperature detection, which confirms the potential of Bi-MAPbBr₃ for constructing the

sensitive thermograph. Figure 1f,g shows normalized absorption spectra in MAPbCl₃ SCs and 10% Bi-MAPbCl₃ SCs at various temperatures. The linear variation of absorption on temperature was also found in Bi-MAPbCl₃ SCs from 303 to 323 K (Figure 1h).

As we demonstrate above, the incorporation of Bi in MAPbX₃ shows temperature-sensitive absorption. Taking the as-fabricated MAPbBr₃ with 10% doping concentration as an example, Figure 2a gives its XRD profile. Figure S2 presents the absorption of Bi-MAPbBr₃ at different doping concentrations. Notably, herein, the doping concentration is the molar ratio of BiX₃ in the precursor. To provide a visual comparison, the data of the intrinsic MAPbBr₃ are presented. As we can see, both MAPbBr₃ and Bi-MAPbBr₃ SCs exhibit the pure cubic phase and high crystal crystallinity. Due to the similar ionic radii of Bi³⁺ (1.03 Å) and Pb²⁺ (1.19 Å), the characteristic peaks indicate a negligible change in the lattice. Similar to MAPbBr₃, the Bi ion can also be doped into MAPbCl₃, and their XRD pattern can be found in Figure S3. Figure 2b displays the PL spectra in MAPbBr₃ and Bi-MAPbBr₃. The incorporation of Bi not only reduces the PL intensity but also shifts the PL peak from 543 to 552 nm. This result indicates that Bi doping can narrow the optical band gap of MAPbBr₃ and enhance the nonradiative recombination.³¹ Additionally, Figure S4 shows the PL spectra of undoped MAPbCl₃ SCs and Bi-MAPbCl₃ SCs. Figure S5 shows the major carrier concentration in Bi-MAPbBr₃, MAPbBr₃ and Bi-MAPbCl₃,

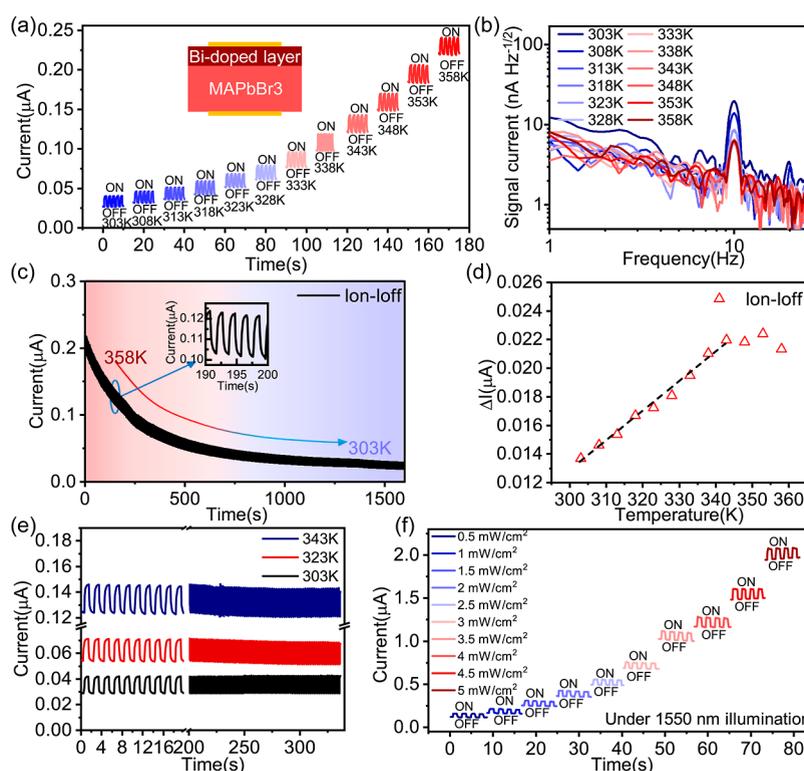


Figure 3. Thermal response properties of the device based on the epitaxial Bi-MAPbBr₃ layer. (a) Photocurrents at 660 nm of the device based on Au/Bi-MAPbBr₃/undoped MAPbBr₃/Au at different temperatures. (b) Signal-to-noise currents measured by the FFT of the device with Bi-MAPbBr₃ layers under -10 V. (c) Difference between I_{ON} and I_{OFF} of the device at different temperatures. (d) Photocurrents at 660 nm of the device with the temperature decreasing from 358 to 303 K. (e) Long-term temperature-sensitive response stability of the device. (f) Temporal photoresponse at 660 nm of device A under 1550 nm IR illumination with different intensities.

MAPbCl₃ from the Hall effect measurement.^{28,29} The incorporation of Bi³⁺ ions can change the majority carrier from holes to electrons and significantly increase their concentration. Figure 2e displays the energy level in MAPbBr₃ and Bi-MAPbBr₃, in which the differences between the Fermi level (E_{F}) and the valence band maximum (E_{VBM}) were estimated from XPS results (Figure 2c), and the E_{F} is extracted from KPFM (Figure 2d).³² The differences of E_{F} and E_{VBM} were measured to be 1.5 and 0.85 eV for Bi-MAPbBr₃ SCs and undoped MAPbBr₃ SCs, respectively, and the Fermi levels of undoped MAPbBr₃ SCs and Bi-MAPbBr₃ SCs are 0.62 and 0.83 eV higher than that of gold, respectively.

Bi doping enhances the carrier recombination, as well as temperature-sensitive absorption of MAPbBr₃. In this case, a p-n junction with homoepitaxial Bi-MAPbBr₃ layers (n-type) on MAPbBr₃ SC (p-type) substrates was designed to optimize the carrier dynamics. Figure 2f shows the two steps of the SPE growth process of homoepitaxial Bi-MAPbBr₃ layers. Step 1: the MAPbBr₃ SC structure was placed into the Bi-MAPbBr₃ precursor solution. Step 2: the extra area was cut off, and the surface was polished. Figure 2g presents its top-view SEM image, in which the Bi-MAPbBr₃ layer shows a smooth and compact surface. The intact surface proves that the Bi-MAPbBr₃ SC is of high quality. It is important to improve the device performance since the published studies indicated that the carrier recombination mainly occurs at the crystal domain surface.^{33,34} The inset in Figure 2g displays the photograph of the as-fabricated thermograph with Bi-MAPbBr₃/MAPbBr₃ SCs and Au electrodes. As we can observe, the visual color at the interface changes from orange in MAPbBr₃ to dark

brown in Bi-MAPbBr₃. Figure 2h provides the cross-section SEM image. It identifies the thicknesses in Bi-MAPbBr₃ and MAPbBr₃ layers as 267 μm and 3 μm , respectively. The thin thickness in Bi-MAPbBr₃ enables the built-in potential to drive its thermal energy generated carriers to the external circuit. The energy-dispersive X-ray spectroscopy (EDX) spectra of the Bi elements in devices are shown in Figure S6. A clear emergence of the Bi element appears in the Bi-MAPbBr₃ SC layer.

Meanwhile, the HRTEM image (Figure 2i) reveals a well-aligned cubic lattice between epitaxial layers without dislocations.³³ This result verifies that the Bi doping will not change the lattice parameters, which coincides with the XRD result. Moreover, the perfect lattice match ensures charge carrier transport at the interface between Bi-MAPbBr₃ and MAPbBr₃ layers.³⁵

Similar to the measurement of temperature-sensitive absorption, a pulsed light at 660 nm (the most absorption-sensitive wavelength) with a modulated on/off switch frequency of 0.5 Hz (10 Hz) was employed. Figure 3a depicts the photoresponse toward 660 nm photons in our thermograph at a -10 V bias under various temperatures.

The temperature of the sample chamber increased from 303 to 358 K with an interval of 5 K. As shown, the dark currents increase with the temperature due to the bolometric effect in the perovskite, as revealed in our previous work.¹⁸ Even so, the photocurrents display an obvious enhancement under 660 nm illumination. To prove the feature that the differences between I_{ON} and I_{OFF} increase linearly with temperature is caused by epitaxial Bi-doped layers, we tested the photocurrents of

Table 1. Comparison of Typical IR Thermal Detectors

device structure	responsivity	D^* (Jones)	sensitivity (%/K)	reference
a-Si _x Ge _y /H	2×10^{-3} A/W	7×10^9	0.043	39
V ₂ O ₅	36 V/W	6×10^5	2.8	40
poly-SiGe	15,000 V/W	8.3×10^8	-1.91	41
metal multiwalled carbon nanotube	0.503 V/W		-0.11	42
PV/CaF ₂ /NiCr/HgTe QD/Ag @290 K	1×10^{-3} A/W	1.2×10^7		43
PV/ITO/HgTeQD/Ag ₂ Te QD/HgCl ₂ /Au @295 K	8×10^{-2} A/W	3×10^8		14
PbSe @4 μm	0.64 A/W	8.36×10^8		44
PbSe/CdSe @4.2 μm	7.5×10^{-2} A/W	1.02×10^9		45
Au/Bi-MAPbBr ₃ /undoped MAPbBr ₃ /Au	2.06×10^{-3} A/W	2.08×10^9	2.11	this work

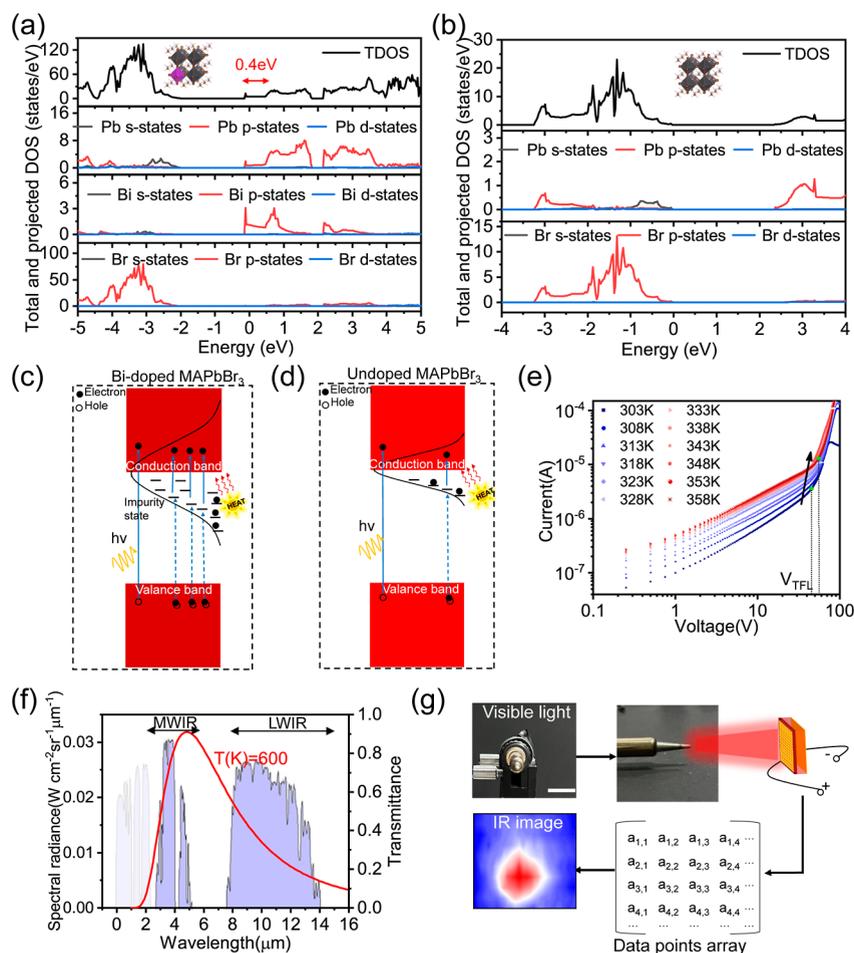


Figure 4. Schematic diagram of the device with the Bi-doping-induced trap state in MAPbX₃. Total and projected DOS diagrams of (a) undoped MAPbBr₃ and (b) MAPbBr₃ with 10% Bi doping. (c,d) Proposed charge carrier excitation process of Bi-MAPbBr₃ and undoped MAPbBr₃, respectively. (e) Trap density measured by the space-charge-limited-current (SCLC) method at various temperatures. Application of the device for IR thermal imaging. (f) Electromagnetic radiation emitted by a soldering iron tip at 600 K. (g) Schematic diagram of the IR thermal imaging setup.

detectors based on Au/undoped MAPbBr₃/Au at various temperatures for comparison, as depicted in Figure S7. There is no obviously enhancement for the photocurrents of the device based only on undoped MAPbBr₃ SCs under 583 nm illumination. Figure 3b illustrates the signal-to-noise ratio measured by the fast Fourier transform (FFT) under 660 nm photon illumination with a modulation frequency of 10 Hz at various temperatures. Clearly, the photocurrents signal peak gradually decreases with the temperature from 19.71 nA Hz^{-1/2} at 303 K to 6.27 nA Hz^{-1/2} at 358 K.³⁶ It confirms the temperature-sensitive electric signal in our thermograph originating from the response of 660 nm photon irradiation.

To extract the precise influence of temperature on photocurrent, the differences between photocurrents (I_{ON}) and dark currents (I_{OFF}) under various temperatures are obtained. It is clearly shown in Figure 3c that the differences between I_{ON} and I_{OFF} increase linearly with temperature from 303 to 343 K, and a saturation emerges at temperatures over 343 K. This phenomenon is the same as that in temperature-sensitive absorption. It is reasonable to ascribe the variation in the differences between I_{ON} and I_{OFF} to the change in the 660 nm photon absorption. Figure 3d displays the differences between I_{ON} and I_{OFF} with temperature continuously decreasing from 358 to 303 K. Figure S8 gives the variation of the photocurrent

difference as the temperature decreases. Especially, during the cooling, the photocurrent difference still presents a linear dependence on temperature, which proves the excellent repeatability of our thermograph. To quantify the temperature sensitivity of our devices, we defined the thermal response current coefficient as

$$k \equiv \frac{\partial(I_{\text{ON}} - I_{\text{OFF}})}{(I_{\text{ON}} - I_{\text{OFF}})\partial T} \quad (2)$$

The sensitivity in our device is estimated to be 2.11% K⁻¹, which is several orders of magnitude larger than that for the temperature coefficient of resistance (TCR) of typical resistance thermometers.³⁷ Moreover, our devices have a high temperature resolution of 0.21 nA K⁻¹ in the linear dynamic range. Figure 3e shows the long-term temporal response in our thermograph toward pulsed 660 nm photons with the power of 2.5 mW cm⁻² under various temperatures. Also, no change in the baseline or photoresponse is observed, which indicates a good temperature tolerance and good repeatability. It is essential to ensure the temperature resolution in thermography.

In general, the requirement in practical applications for thermographs is the capacity for remote sensing. To check the potential of our device for noncontact real-time thermography, the 1550 nm photon is adopted to simulate the thermal radiation. Figure 3f shows the temporal response of our thermograph to 660 nm photons under 1550 nm irradiation with different intensities. Our device absorbs the IR photon and converts it into thermal energy, which subsequently affects the photocurrent resulting from 660 nm photon irradiation. Thus, the photocurrent difference increases as the IR intensity increases. To better describe this performance, the responsivity (*R*) and detectivity (*D*^{*}) were calculated from the photocurrents. *R* is defined as

$$R = \frac{I_{\text{ON}} - I_{\text{OFF}}}{P_{\text{light}} \times S} \quad (3)$$

where *S* is the effective area and *P*_{light} is the light intensity. *D*^{*} is simplified as

$$D^* = R \sqrt{\left(\frac{s}{2qI_{\text{OFF}}}\right)} \quad (4)$$

in which *q* is the electron charge (1.6 × 10⁻¹⁹ C).³⁸ Figure S9 gives the calculated *R* and *D*^{*} at different intensities. The highest *R* and *D*^{*} in our thermograph under 0.5 mW cm⁻² of 1550 nm illumination were calculated to be 2.06 mA W⁻¹ and 2.08 × 10⁹ Jones, respectively. To better understand the advances, Table 1 gives the comparison of the performance between our thermograph and the typical IR thermal detectors.

The importance of Bi incorporation is underscored by the results from temperature-sensitive absorption and visible photon detection. To further clarify the role of Bi doping, we next investigate the density of states (DOS) in Bi-MAPbBr₃ and MAPbBr₃ with the assistance of DFT calculation.²³ Figure 4a,b presents the calculation result. Due to the characteristics of the electron donor, Bi ions make a significant contribution to the DOS. As we can see, the incorporation of Bi ions brings a non-negligible trap state distribution in the forbidden band, and the edge of the addition state is ~0.4 eV to the conduction band maximum (CBM) of MAPbBr₃. It is worth noting that the distribution of extra states is continuous and extends to the

CBM. To better understand the effect of Bi doping on the temperature-sensitive absorption, Figure 4c,d shows the proposed charge carrier excitation process. The additional trap states enable the acceptance of carriers pumped by the absorption of visible photons with energy lower than the band gap. The overlap between trap states and the conduction band in Bi-MAPbBr₃ facilitates some trapped charge carriers jumping to the conduction band under thermal energy excitation. The left trap state is empty, and it enhances the subsequent absorption (Figure 3a). This dynamic process results in temperature-sensitive absorption. To corroborate this process, the trap density (*n*_{traps}) under various temperatures was estimated using the SCLC method, as shown in Figure 4e. The formula is

$$n_{\text{traps}} = \frac{2\varepsilon V_{\text{TFL}}}{qL^2} \quad (5)$$

where ε is relative dielectric constant, *V*_{TFL} is the trap-filled limit voltage at which all injected charge carriers are used to fill all traps, *q* is the electronic charge, and *L* is the distance between the opposite electrodes.⁴⁶ As expected, with the temperature increasing from 303 to 358 K, the *V*_{TFL} increases from 45.5 to 56.0 V, leading to the calculated *n*_{traps} increasing from 2.85 × 10¹¹ cm⁻³ to 3.51 × 10¹¹ cm⁻³. Obviously, the trap/detrapping processing in the additional states originating from Bi doping plays an indispensable role in realizing this temperature-sensitive functionality.

To further verify the potential of our thermograph in practical application, the noncontact real-time thermal imaging was conducted with an electric soldering iron tip as the sensing target. Under work conditions, the temperature on the electric soldering iron tip approaches 600 K. According to the blackbody radiation law, its radiation wavelength can be calculated with the following equation

$$u_{\nu}(\nu, T) = \frac{8\pi h\nu^3}{c^3} \frac{1}{e^{\frac{h\nu}{kT}} - 1} \quad (6)$$

in which *h* is the Planck constant, ν is the frequency, *c* is the speed of light, and *k* and *T* are the Boltzmann constant and temperature, respectively.¹⁰ As shown in Figure 4f, the wavelength of IR radiation from the soldering iron tip mainly localizes in the atmospheric window (3–5 and 8–13 μm),¹¹ which endows the remote thermal imaging. Figure 4g illustrates the schematic diagram of a homemade thermal imaging setup, in which the distance between our thermograph and the electric soldering iron tip is ~30 cm.

Clearly, the obtained patterns perfectly reflect the temperature distribution around the electric soldering iron tip. The gradient change in the imaging also further identifies the advantage of high temperature resolution in our thermograph. According to the practical condition for thermal imaging of the soldering iron tip, in which the thermal energy density on each pixel is ~8.2 × 10⁻² W m⁻², it is reasonable to deduce that even without the assistance of an optical system, our thermograph can be used as a noncontact thermometer.

CONCLUSIONS

As conclusions, we discovered that the Bi-MAPbX₃ SCs have a linear temperature dependence of absorption due to the trap states induced by Bi dopants. According to these temperature-induced changes in the absorption, we fabricated a thermo-

graph with p–n junctions based on the homoepitaxial Bi-MAPbBr₃ (n-type) on the intrinsic MAPbBr₃ (p-type) substrates. The differences between I_{ON} and I_{OFF} by sensing visible photons is successfully applied to characterize the temperature and thermal imaging, in which a high responsivity of 2.06 mA W⁻¹, a high detectivity of 2.08×10^9 Jones, and a high sensitivity of 2.11% K⁻¹ are achieved. We believe that the detection of visible photons in our thermograph will enable its integration with the conventional optical system for imaging with high spatial resolution and thermal resolution.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsami.3c13305>.

The absorption of Bi-MAPbBr₃ with different doping concentrations; XRD patterns of undoped MAPbCl₃ and Bi-MAPbCl₃, respectively; Hall effect of undoped and 10% Bi-doped SCs to measure the majority carrier density and type; EDX results for SC devices with epitaxial Bi-doped layers; the variation of the photocurrent difference with the decrease in the temperature; and calculated responsivity (R) and detectivity (D^*) of the device (PDF)

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Author Contributions

Y.X. and Y.P. grew the perovskite single crystals. Y.X. and X.W. conducted the epitaxial experiments. Y.X. and X.W. obtained the measurements. L.S. performed the DFT calculations. Y.X., X.W., W.L., and X.X. analyzed these results. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS

MA, methylammonium; IR, infrared; SCs, single crystals; XRD, X-ray diffraction; PL, photoluminescence; ITC, inverse temperature crystallization; SEM, scanning electron micros-

copy; HRTEM, high-resolution transmission electron microscopy; FFT, fast Fourier transform; EDX, energy-dispersive X-ray spectroscopy; KPFM, Kelvin probe force microscopy; TCR, temperature coefficient of resistance; DFT, density functional theory; CBM, conduction band maximum; SCLC, space-charge-limited-current

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